



TECHNICAL PAPER

FLUE GAS CLEANING DURING THE TORREFACTION PROCESS OF LOW VALUE BIOMASS CONTAINING RESIDUAL FLOWS

Considerations for the method chosen and description of the flue gas cleaning installation.

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ABSTRACT

When using low value biomass containing residual flows as feedstock for the torrefaction process the increased chlorine and sulphur content of these flows will be partly present in the process-torr gas to that extent it will give rise to emission to open air problems (not meeting the emission standards).

Investigations showed, that cleaning of the process gas only can be done in an effective and controlled way after incinerating the process torr-gas.

literature research and consultation with potential suppliers of flue gas cleaning installations indicated, that a dry flue gas cleaning with addition of Ca(OH)_2 or NaHCO_3 is technically and economically the most favourable.

CONTEXT OF THE PAPER

This technical paper is part of a number of articles that describe the use of low-value biomass-containing residual flows as a raw material for the torrefaction process, whereby the torrefied product (bio-carbon) could replace fossil black and brown coal in various types of application.

FLUE GAS CLEANING TORREFACTION PROCESS

INTRODUCTION

Exposing various biomass containing flows to a torrefaction process (heating up to 300 °C in an oxygen low environment), the properties of these flows will be changing in such a way these can be used as a fossil coal substitute for energy production and chemical industrial processes (like syngas production and steel production).

For business economic and sustainable reasons biomass containing flows are used, which cannot be used in their original form for other high-quality applications.

Unfortunately, these biomass containing flows mostly show an increased chlorine content (max. 2 w.-%) and / or sulphur content (max. 0,5 w.-%). During torrefaction the present chlorine and sulphur move partly to the process-torr gas and partly these elements stay behind in the torrefied product.

The process torr-gas is burned during the torrefaction process and flue gas will come into existence with HCl and SO₂ content far above the allowed emission standard to open air.

By that it is necessary to remove sufficiently chlorine and sulphur from the process torr-gas or the flue gas before emission to open air takes place (gas cleaning).

GAS CLEANING

About 50 % up to 90 % of the chlorine, present in the biomass feedstock moves during torrefaction to the process torr-gas. Investigations have shown, that about 40 % of the chlorine, present in the process torr-gas, is organically bound. The remaining 60 % is present as inorganic HCl.

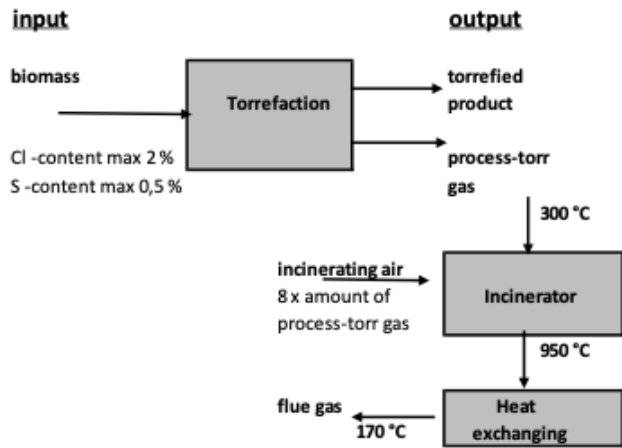
Only the present inorganic HCl can be removed with the common additives, like Ca(OH)₂ or NaHCO₃. By that it is not possible to remove chlorine in a sufficient way from the process torr-gas. After burning of the process torr-gas all chlorine will be present in the flue gas as inorganic HCl and can be removed almost 90 % by Ca(OH)₂ or NaHCO₃ at temperatures between 100 °C and 300 °C.

About 20 % up to 60 % of the sulphur, present in the biomass feedstock moves during torrefaction to the process torr-gas. Investigations done by ECN (commissioned by Torr-Coal Technology) have shown, that about 58 % of the sulphur in the process-torr gas will be present as SO₂, 38 % as H₂S and 4 % as COS (carbonyl sulphide). It is possible to remove at about 300 °C with certain common additives SO₂ and H₂S from the process torr-gas.

After burning of the process torr-gas all sulphur will be present in the flue gas as SO₂ and SO₃. These compounds can be removed for a large part by Ca(OH)₂ or NaHCO₃ at temperatures between 100 °C and 300 °C

Besides the fact chlorine cannot be removed in a sufficient degree from the process torr-gas there is another big disadvantage, if the removal of chlorine and sulphur will be done from the process torr-gas. This gas contains, due to the torrefaction process, an amount of tars, which may have a disturbing effect during the removal of chlorine and sulphur from the process torr-gas.

SUMMARIZED:



One remark about the flue gas temperature. Looking to

- melting points of salts formed,
- reactivity of the additive with HCl, HF, SO₂ and SO₃,
- acid dewpoints (SO₃ most critical: about 140 °C)
- temperature resistance of the filter bags,

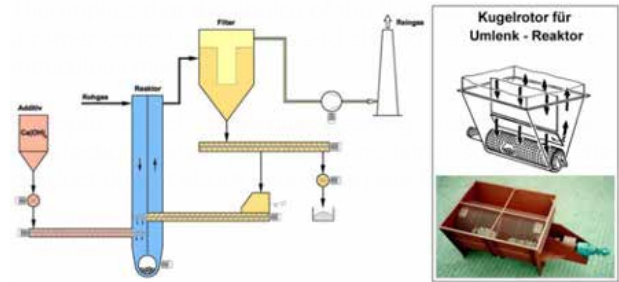


Fig. 1: sketch flue gas cleaning installation

Distribution Cl and S: input to output

	torrefied product		process-torr gas	
	min.	max.	min.	max.
Cl	10%	50%	50%	90%
S	40%	80%	20%	20%

Appearance of Cl and S in process-torr gas (approx.)

Cl		S	
organical	40%	SO ₂ /SO ₃	58%
inorganical	60%	H ₂ S	38%
		COS	4%

Appearance of Cl and S in flue gas

Cl		S	
organical	0%	SO ₂ /SO ₃	100%
inorganical	100%	H ₂ S	0%
		COS	0%

The best way to prevent unacceptable emissions to open air is therefore to clean the flue gas (this means gas cleaning after incinerating the process torr gas) by adding to the flue gas dry Ca(OH)₂ or NaHCO₃ at a flue gas temperature of about 200 °C. This is a proven flue gas cleaning technology.

In other words, this flue gas cleaning installation is technically and process-based the most confident to realize the set emission to open air goals.

Flue gas cleaning installation consists of: a binder storage silo; transport system binder to reactor; a reactor (mixing binder / flue gas); flat-bag filter to separate used binder and cleaned flue gas; back and discharge of used binder; flue gas fan. Processing about 20.000 Nm³/hour flue gas total needed electrical power (installed) will be about 160 kW. Binder consumption depends on the amount of HCl, SO₂ and SO₃.

CONCLUSION

Low value biomass containing rest streams, like agricultural residues, wood from park and forest maintenance (branches), demolition wood and separated household waste (SRF), can be used as a raw material for the torrefaction process, whereby the torrefied product (bio-carbon) could replace fossil black and brown coal in various types of application.

Some of these raw materials will show an increased content chlorine (max. 2 w.-%) and / or sulphur content (max. 0,5 w.-%) and by that these elements will be partly present in the process-torr gas (chlorine up to 90 % of original quantity in the raw material; sulphur up to 60 % of original quantity in the raw material) to that extent it will give rise to emission to open air problems (not meeting the emission standards).

It will be for sure an additional process is needed in that case to meet the prescribed emission to open air standards.

Investigations showed, that cleaning of the process torr-gas only can be done in an effective and controlled way after incinerating the process torr-gas.

Literature research and consultation with potential suppliers of flue gas cleaning installations and additives indicated, that a dry flue gas cleaning with addition of $\text{Ca}(\text{OH})_2$ or NaHCO_3 is technically and economically the most favourable.

This gas cleaning system has been installed at the industrial torrefaction production installation in Dilsen-Stokkem (Belgium) end 2020. By that it will be possible to meet the prescribed emission standards, if biomass containing low-value residual flows with increased chlorine and sulphur content are used for torrefaction on industrial scale.

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